

## Research Article

# An improved method for the radiosynthesis of [ $^{11}\text{C}$ ]d-threo-methylphenidate

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## Summary

An improved method to remove the 2-nitrophenylsulfenyl protecting group in the conventional radiosynthesis of the PET radiotracer [ $^{11}\text{C}$ ]d-threo-methylphenidate was developed. The method uses the nonvolatile reagents sulfuric acid and L-cysteine. In addition it was demonstrated that solid phase extraction (SPE, C18 and ion exchange) can be used to purify the final product instead of semi-preparative HPLC. Copyright © 2006 John Wiley & Sons, Ltd.

**Key Words:** [ $^{11}\text{C}$ ]methylphenidate; dopamine transporter; SPE purification

## Introduction

Changes in the dopamine system are observed in a number of diseases such as Parkinson's disease, schizophrenia and substance abuse.<sup>1</sup> The conventional method for the radiosynthesis of a radiotracer for the dopamine transporter, [ $^{11}\text{C}$ ]d-threo-methylphenidate ([ $^{11}\text{C}$ ]MP),<sup>2</sup> producing adequate quantity and quality of the radiotracer, has been successfully used for over a decade. Our research was aimed at making the radiosynthesis safer for radiochemists, simpler, faster and more reliable. The original reference<sup>2</sup> and our research suggests that the synthesis of [ $^{11}\text{C}$ ]MP by direct  $^{11}\text{C}$ -iodomethylation of ritalinic acid is impractical. Alternative  $^{11}\text{C}$ -methylation agents such as

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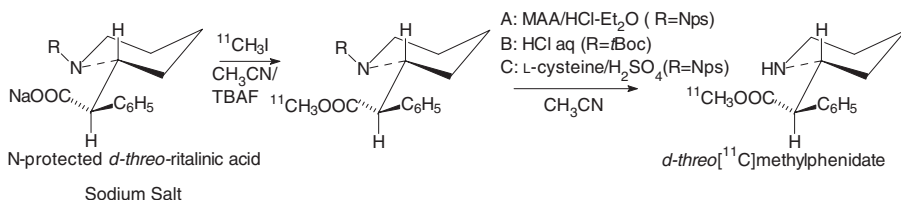
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Contract/grant sponsor: TRIUMF

[ $^{11}\text{C}$ ]methyl triflate<sup>3</sup> and [ $^{11}\text{C}$ ]CH<sub>3</sub>OH/BF<sub>3</sub>-Et<sub>2</sub>O<sup>4</sup> were also unsuccessfully tried.

The use of a *t*-Boc-protecting group-containing precursor and dilute aqueous HCl for deprotection was proven feasible but low yielding. However, substitution of HCl-diethyl ether with concentrated H<sub>2</sub>SO<sub>4</sub> and mercaptoacetic acid with L-cysteine, while retaining the protecting 2-nitrophenylsulfenyl-(Nps) group, worked well and eliminated the disadvantages associated with the volatile reagents. To accelerate [ $^{11}\text{C}$ ]MP purification, a previously developed<sup>5</sup> solid phase extraction (SPE) system utilizing two Sep-Paks was considered for the final product purification as an alternative to HPLC.

## Experimental

Chemicals were purchased from Aldrich or Across. [ $^{11}\text{C}$ ]CH<sub>3</sub>I was produced using gas-phase method from in-target generated [ $^{11}\text{C}$ ]CH<sub>4</sub>. The final solution of radiotracer was filtered through sterile Acrodisc<sup>TM</sup> filter (0.22  $\mu\text{m}$ , 25 mm, Pall Corp.). [ $^{11}\text{C}$ ]MP quality control was accomplished using Waters HPLC with Nova-Pak, C18, 5  $\mu\text{m}$ , 150  $\times$  4.6 mm column; 0.17 M ammonium formate – 30% acetonitrile eluent at 1.5 ml/min; UV 225 nm; *t*<sub>R</sub> 2.5 min.



**Figure 1. Radiosynthesis of [ $^{11}\text{C}$ ]d-threo-methylphenidate; MAA-mercaptoacetic acid**

### [ $^{11}\text{C}$ ]MP radiosynthesis<sup>2</sup>

[ $^{11}\text{C}$ ]CH<sub>3</sub>I in a stream of helium was bubbled through the cooled (aluminum cylinder, 5–10°C) solution of MP precursor [*d*-threo-*N*-(2-Nitrophenylsulfenyl) ritalinic acid sodium salt, 0.3 mg, 0.8  $\mu\text{mol}$ , JML Biopharm, Vancouver, BC, Canada] and tetrabutylammonium fluoride (1.3  $\mu\text{mol}$ ) in anhydrous acetonitrile (0.4 ml). The mixture was heated (oil bath, 80°C, 5 min), and a solution of 5.5–8.5 mg (45–70  $\mu\text{mol}$ ) of L-cysteine and 17  $\mu\text{l}$  (340  $\mu\text{mol}$ ) of concentrated sulfuric acid in 0.4 ml of acetonitrile was added via a remote addition line. After stirring at room temperature for 3 min, the reaction was quenched with 3 ml of saturated sodium tetraborate. The solution was loaded onto C18 Light Sep-Pak (Waters), and the cartridge was washed with water (4 ml) (Figure 1).

### HPLC Method

[ $^{11}\text{C}$ ]MP was eluted (methanol, 2 ml) from the Sep-Pak into the 2-ml injector loop. The product was purified by HPLC (Phenomenex, Partisil, 10  $\mu\text{m}$ , 250  $\times$  9.4 mm column; 0.17 M ammonium formate–25% acetonitrile eluent at 5 ml/min; UV 254 nm;  $t_{\text{R}}$  10–12 min). Collected product fraction was rotary evaporated (60–70°C); ethanol (5 ml) was added and evaporated to remove traces of acetonitrile, and the radiotracer was reconstituted in saline (10 ml). The decay corrected radiochemical yield with respect to [ $^{11}\text{C}$ ]CH $_3$ I was  $50 \pm 15\%$  ( $n = 6$ ) 45 min after EOB; radiochemical purity was  $>99\%$ ; and EOS specific radioactivity,  $8 \pm 5$  ( $n = 6$ ) Ci/ $\mu\text{mol}$ .

### SPE Method

The solution of product, eluted from C18 Sep-Pak with ethanol (2 ml), was mixed with water (20 ml) and passed through Waters CM Sep-Pak. After the cartridge was washed with water ( $2 \times 5$  ml), [ $^{11}\text{C}$ ]MP was eluted with saline (10 ml). The decay corrected radiochemical yield with respect to [ $^{11}\text{C}$ ]CH $_3$ I was  $45 \pm 15\%$  ( $n = 4$ ) after 45 min of radiosynthesis; radiochemical purity was  $>99\%$ , and EOS specific radioactivity was  $1.8 \pm 0.5$  ( $n = 4$ ) Ci/ $\mu\text{mol}$ .

## Results and discussion

Although ritalinic acid has two potential sites for  $^{11}\text{C}$ -methylation, it has been shown that chemoselective  $^{11}\text{C}$ -methylation can, in principle, be achieved by changing the reaction solvent or  $^{11}\text{C}$ -methylating agent.<sup>4,6–8</sup> In our case, however, no formation of [ $^{11}\text{C}$ ]MP was observed in the reaction of ritalinic acid with carbon-11- methyl iodide, methyl triflate and CH $_3$ OH/BF $_3$ -Et $_2$ O. In the conventional radiosynthesis a mixture of HCl-diethyl ether and mercaptoacetic acids is required for deprotection of the Nps- group;<sup>2</sup> both reagents are volatile irritants, and HCl-diethyl ether is unstable upon storage. The use of *t*-Boc as a protecting group in the radiosynthesis of [ $^{11}\text{C}$ ]MP afforded the use of a milder deprotection system (1 M aqueous HCl). However, this resulted in a 30% lower radiochemical yield compared to the original radiosynthesis.

Another proposed improvement to the radiosynthesis was the development of a better method to remove the Nps-protecting group in the conventional radiosynthesis. These attempts have proven successful; a new deprotection system using a solution of nonvolatile L-cysteine and sulfuric acids in acetonitrile avoids the disadvantages associated with HCl-diethyl ether/mercaptoacetic acid without sacrificing radiochemical yield.

Due to the unlikely formation of *l*-erythro-[ $^{11}\text{C}$ ]MP, the final product can be purified using the Sep-Pak method;<sup>5</sup> completion of QC, however, would be required prior to administration. Meanwhile, preparative HPLC indicates, in

the course of the radiosynthesis, whether significant amount of *l*-erythro- $^{11}\text{C}$ MP is formed and separates these two diastereomers. Another disadvantage of the Sep-Pak radiosynthesis is a requirement for a dedicated setup as opposed to the use of universally applicable radioHPLC with multi-column/eluent selectors. An initial believe in a shorter time of Sep-Pak radiosynthesis was deceptive, especially considering the added time for QC release.

## Conclusions

Use of the L-cysteine/ $\text{H}_2\text{SO}_4$  deprotection system with the conventional radiosynthesis proved to be advantageous compared to the mercaptoacetic acid/HCl-diethyl ether method. Our other experiments showed that a one-step conversion of ritalinic acid into  $^{11}\text{C}$ methylphenidate is not practical, while the search for an alternative protective group to the currently used Nps-group is nontrivial.

## Acknowledgements

We are thankful to Canadian Institutes of Health Research and TRIUMF for financial support, P. Piccioni for operating the cyclotron and T. J. Ruth for reviewing our manuscript.

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